

Palestine Technical University-Kadoorie Course Syllabus



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Faculty:		<u> </u>				
Course Title:	General Chemistry (I)	Course Number:	15050101			
Year:	2016/2017	Semester:				
Department:	Chemistry	Designation:				
Prerequisite(s):	None	r				
Instructor:	Dr. Mansour Ararawi, Dr. Sameer Bsharat,					
	Ms. Nour Abdul Rahman, and Dr. Lamees Z.					
	Majjad					
Instructor's e-mail:	lameesmajjad@yahoo.com., mansourararawi@yahoo.com., s.bsharat@ptuk.ed					
	nour.nayef@najah.edu					
Office Hours:		-				
Class Time:		Class Room:				
Course description:	This course is designed to give students a working knowledge of the most important					
description.	chemical principles as the foundation for study of more advanced topics such as					
	chemical analysis, applied chemistry, inorganic and organic chemistry.					
Textbook(s):	General Chemistry, Ebbing and Gammon 8 th edition					
Other required						
material						
(References): Course objectives:	On successful completion of this course the student will be able to					
	1- Demonstrate an understanding of basic chemical nomenclature and formulas.					
	2- Explain concepts of basic atomic theory and relate the theory to the periodi table.					
	3- Write chemical reactions and solve problems involving chemical stoichiometry.					
	4- Describe the nature of aqueous solutions and reactions occurring in aqueous					
	solution.					
	5- To apply the Ideal Gas Law equation, Avogadro's Law and Dalton's Law.					
	6- Apply concepts of thermochemistry to physical and chemical changes.					
	7- Describe the electronic structure of atoms and relate the electronic structure to					
	atomic properties.8- Demonstrate an understanding of chemical bonding and its application to molecular structure.					
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Topics covered and Calendar:	T !		Weeks or			
	Topics		number			
	Chemistry and measurements		of hours			
	Chemistry and measurements					
	Atoms, Molecules, and Ions					
			5			
	Calculations with Chemical Formulas and Equa	ations	4			
	Chemical Reactions	ations				
		utions	4			
	Chemical Reactions	ations	4 6			

Prepared by: Dr. Lamees Z. Majjad			Date: 15 February 2017		
	3- All mobiles must be switched off during class				
CIASS FUILIES	marked absent				
General Notes: Class Policies	1-University regulation Regarding absentees will be Applied 2-Names will be read at the beginning of the class and anyone coming after that will be				
Conoral Notao-	Final Exam	(40 Points)	Will be announced by the registrar		
	Semester works	(00 Points)			
	Second Exam	(30 Points)			
Grading Plan:	First Exam	(30 Points)			
	Molecular Geometry and Chemical Bonding Theory 3			3	
	Ionic and Covalent Bonding			5	
	Electron Configur	ation and Per	iodicity	3	